

भारत सरकार :: अंतरिक्ष विभाग Government of India :: Dept. of Space

सतीश धवन अंतरिक्ष केंद्र शार, श्रीहरिकोटा

भारतीय अंतरिक्ष अनुसंधान संगठन Indian Space Research Organisation



SATISH DHAWAN SPACE CENTRE SHAR Stilbertikota

Section	n : Chemical Engineering BE and B Tech	
Q.1	Unit of mass velocity is	
	 A. Kg / hour B. Kg / m² hour C. Kg/m² D. Kg/m³ hour 	
Ans	X 1. A	
	✓ 2. B	
	X 3. C	
	★ 4. D	
		Question ID : 5834939933
Q.2	The second law of thermodynamics states that	
	A. Energy change of a system undergoing any reversible process	is zero
	B. It is not possible to transfer heat from lower temperature to hig	ther temperature
	C. Total energy of the system and the surroundings remains const	ant.
	D. None of these	
Ans	★ 1. A	
	✓ 2. B	
	X 3. C	
	X 4. D	
		Question ID : 5834939925
Q.3	Buckingham ∏- theorm states that in any physical problem including 'm' dimensions, the quantities can be arranged into indepe	
	parameter	
	A. m	
	B. n	
	C. n-m	
	D. n/m	
Ans	X 1. A	
	★ 2. B	
	√ 3. C	
	★ 4. D	
		Question ID : 5834939936

Q.4	The temperature at which, a real gas obeys the ideal gas law, over a	a range of pressure
	is called	
	A. Critical temperature	
	B. Inversion temperature	
	C. Reduced temperature	
	D. Boyle temperature	
Ans	★ 1. A	
	X 2. B	
	X 3. C	
	✓ 4. D	
		Question ID : 5834939926
Q.5	Toluene-di-isocyanate is used in making	
	A. Dyestuffs	
	B. Flexible polyurethane foams	
	C. Pharmaceutical products	
	D. All of the above	
Ans	※ 1. A	
	✓ 2. B	
	X 3. C	
	X 4. D	
		Question ID : 5834939958
Q.6	The ratio of the actual mesh dimension of Taylor series to that of the nex	ct smaller screen
	A. 2	
	B. $\sqrt{2}$	
	C. 1.5	
	D. $\sqrt{3}$	
Ans	★ 1. A	
	✓ 2. B	
	X 3. C	
	★ 4. D	
		Question ID : 5834939940

Q.7	Two pipes of same length and diameters d & 2d respectively are con	nnected in series. The
	diameter of an equivalent pipe of same length is	
	A. Less than d	
	B. Between d & 1.5dC. Between 1.5d & 2d	
	D. Greater than 2d	
Ans	✓ 1. A	
Alis		
	X 2. B	
	X 3. C	
	★ 4. D	
		Question ID : 5834939938
		2.00.00.00
Q.8	Which of the following is not an elastomer?	
	A. Polyisoprene	
	B. Neoprene	
	C. Nitrile-butadiene	
	D. None of these	
Ans	★ 1. A	
	X 2. B	
	★ 3.0	
	✓ 4. D	
	4. 0	
		Question ID : 5834939969
Q.9	Extraction of Coffee from its seed is done by	
	A. Liquid-liquid Extraction	
	B. Leaching	
	C. Extractive distillation	
	D. Steam distillation	
Ans	★ 1. A	
	✓ 2. B	
	X 3. C	
	★ 4.D	
	Mars.	
		Question ID : 5834939944
Q.10		B B 1
Q.10	The reduction ratio for grinders is defined as $(D_f = Feed particle Diameter)$	meter; $D_p = Product$
	particle diameter)	
	$A. D_f \! / D_p$	
	$B. D_p/D_f$	
	$C. D_f - D_p$	
	$\mathrm{D.D_f} * \mathrm{D_p}$	
Ans		
1	✓ 1. A	
	★ 2. B	
	★ 2. B	
	★ 2. B ★ 3. C	Question ID : 5834939941

Q.11 The elementary liquid phase decomposition reaction $A \stackrel{k}{\to} 2B$ is to be carried out in a

CSTR. The design equation is

- A. $k\tau = \frac{X_A}{(1-X_A)}$
- B. $k\tau = \frac{X_A (1+X_A)}{(1-X_A)}$
- C. $k\tau = \frac{X_A}{(1-X_A)^2}$
- D. $k\tau C_{A0} = \frac{x_A}{(1+X_A)^2}$

Ans 🚀 1. A

- X 2. B
- **X** 3. C
- X 4. D

Question ID: 5834939953

Q.12 The 2^{nd} order system with transfer function $4/(S^2+2S+4)$ has a damping ratio of

- A. 2
- B. 0.5
- C. 1
- D. 4

Ans X 1. A

- 🥠 2. B
 - X 3. C
 - 🗙 4. D

Question ID: 5834939932

Q.13 2 moles of He is mixed with 2g of H_2 . The molar heat capacity at constant pressure for

the mixture is

- A. 17R/6
- B. 11R/6
- C. 4R
- D. 3R/2

Ans 🚀 1. A

- X 2. B
- X 3. C
- X 4. D

Q.14	A force digital transducer, measures the pressure range of $0-200\mathrm{M}$	N with a resolution
	of 0.1 % of the full scale. The smallest change it can measure is	
	A. 0.2 N	
	B. 0.4 N	
	C. 0.5 N	
	D. 1.0 N	
Ans	✓ 1. A	
	X 2. B	
	※ 3. C	
	★ 4. D	
	N =	
		Question ID : 5834939931
Q.15	A particle 'A' of diameter 10 μ settles in an oil of specific gravity 0.9 and	viscosity of 10 Poise
	under Stokes Law. A particle 'B' of diameter 20 μ settling in the same of	
	velocity,	C
	A. Same as that of A	
	B. 1/4 th that of A	
	C. Twice as that of A	
	D. 4 times as that of A	
Ans	※ 1. A	
	✓ 2. B	
	※ 3. C	
	★ 4. D	
		Question ID : 5834939970
Q.16	Poly Vinyl Chloride is a	
	A. Thermosetting material	
	B. Thermoplastic material	
	C. Fibrous material	
	D. Chemically active material	
Ans	★ 1. A	
	✓ 2. B	
	X 3. C	
	★ 4. D	
		Question ID : 5834939957
Q.17	Soap solution cannot be used with hard water because	
	A. Hard water contains sulphate	
	B. They form insoluble calcium soap with precipitate	
	C. They attract back, the removed dust	
A	D. None of these	
Ans	★ 1. A	
	√ 2. B	
	X 3. C	
	★ 4. D	
		0
		Question ID : 5834939971

Q.18	The weight fraction of methanol in an aqueous solution is 0.64. The	mole fraction of
	methanol X_m satisfies	
	A. $X_m < 0.5$	
	B. $X_m = 0.5$	
	C. $0.5 < X_m < 0.64$	
	D. $X_m \ge 0.64$	
Ans	★ 1. A	
	✓ 2. B	
	X 3. C	
	★ 4. D	
		Question ID : 5834939964
Q.19	Bracket supports are the most suitable for supporting vessel	ls
	A. Thick walled vertical	
	B. Horizontal	
	C. Thin spherical	
	D. None of these	
Ans	✓ 1. A	
	X 2. B	
	★ 3. C	
	★ 4. D	
		O ID
		Question ID : 5834939967
Q.20	The testing pressure of storage tanks and pressure vessels designed as per Indi	ian standard codes
	should be about times the design pressure.	
	A. 1.5 to 2	
	B. 3 to 4	
	C. 4 to 5	
	D. >5	
Ans	✓ 1. A	
	X 2. B	
	X 3. C	
	★ 4. D	
		Question ID : 5834939968

Q.21	A 4 KW, 20 litre water heater is switched ON for 10 min. The heat capacity for water		
	is 4 KJ/Kg-K. Assuming total electrical energy has gone into heating the water, the		
	increase of water temperature is		
	A. 15 °C		
	B. 20 °C		
	C. 26 °C		
	D. 30 °C		
Ans	★ 1. A		
	★ 2. B		
	★ 3. C		
	✓ 4. D		
	Question ID : 5834939929		
Q.22	Catalyst employed for oxidation of ammonia to nitric oxide is		
	A. γ alumina		
	B. Iron		
	C. Platinum – Rhodium		
	D. Nickel		
Ans	★ 1. A		
	X 2. B		
	✓ 3. C		
	★ 4. D		
	Question ID : 5834939959		
Q.23	The most suitable equipment for removing the fine dust particle ($< 1 \mu$ diameter) from air below		
	its dew point will be a/an		
	A. Bag-filter		
	B. Electrostatic precipitator		
	C. Cyclone separator		
	D. Wet scrubber		
Ans	★ 1. A		
	✓ 2. B		
	X 3. C		
	★ 4. D		
	Question ID : 5834939942		

Q.24 Newton's law of viscosity relates to

- A. Shear stress and Viscosity
- B. Velocity gradient and pressure intensity
- C. Shear stress and rate of angular deformation in a fluid
- D. Pressure gradient and rate of angular deformation

Ans X 1. A

- X 2. B
- √ 3. C
- X 4. D

Question ID: 5834939934

Q.25

A reaction of type $A \xrightarrow{1 \text{ st order}, K_1} B \xrightarrow{1 \text{ st order } K_2} C$ conducted in a batch reactor is characterized by

- A. Maximum concentration of B, with time
- B. Maximum concentration of C, with time
- C. Both (A) and (B)
- D. Neither (A) nor (B)

Ans 🚀 1. A

- X 2. B
- X 3. C
- X 4. D

Question ID: 5834939952

Q.26 The head loss due to sudden expansion is

- A. $(V_1^2 V_2^2) / 2 g_c$
- B. $(V_1 V_2)^2 / 2 g_c$
- C. $(V_1 V_2) / 2 g_c$
- D. $(V_1^2 V_2^2) / g_c$

Ans X 1. A

- ✓ 2. B
- X 3. C
- X 4. D

- **Q.27** If two capillary tubes of diameter 0.5 mm and 1 mm are dipped in a pot containing mercury, then rise of mercury is
 - A. Same in both the tubes
 - B. Greater in 1 mm diameter tube
 - C. Greater in 0.5 mm diameter tube
 - D. Zero in both the tubes
- Ans X 1. A
 - 🗶 2. B
 - √ 3. C
 - X 4. D

Question ID: 5834939935

- **Q.28** The conversion (X_A) for a 2nd order irreversible reaction (Constant volume) $A \stackrel{k1}{\to} B$ in batch mode is given by
 - A. $\frac{1}{1+K_1C_{A0}}$
 - B. $\frac{K_1C_{A0} t}{1 + K_1C_{A0} t}$
 - C. $\frac{(K_1 C_{A0} \ t)^2}{1 + K_1 C_{A0} \ t}$
 - D. $\frac{K_1C_{A0} t}{(1+K_1C_{A0} t)^2}$
- Ans X 1. A
 - ✓ 2. B
 - **X** 3. C
 - X 4. D

Question ID: 5834939955

- Q.29 A dam of width 50 m is used to hold water in a reservoir. If the water height is 10 m from the bottom of the dam, what is the total force (F) acting on the dam due to the water? (Assume $g = 10 \text{ m/sec}^2$ and the fluid density is 1000 kg/m^3)
 - A. $F = 12.5 * 10^6 N$
 - B. $F = 25 * 10^6 \text{ N}$
 - C. $F = 50 * 10^6 \text{ N}$
 - D. $F = 5 * 10^6 \text{ N}$
- Ans X 1. A
 - A 1171
 - **√** 2. B
 - X 3. C
 - 🗙 4. D

Q.30 The overall heat transfer co-efficient (U) for a composite wall of thickness x_1 , x_2 , x_3 and corresponding conductivities $k_1 \, k_2, k_3$ is given by

- A. $1/U = k_1/x_1 + k_2/x_2 + k_3/x_3$
- B. U = $k_1/x_1 + k_2/x_2 + k_3/x_3$
- C. $1/U = x_1/k_1 + x_2/k_2 + x_3/k_3$
- D. U = $x_1/k_1 + x_2/k_2 + x_3/k_3$

Ans X 1. A

- X 2. B
- √ 3. C
- X 4. D

Question ID: 5834939924

- Q.31 Component A is diffusing in a medium B. The flux NA relative to a stationary point is equal to the flux due to molecular diffusion, if
 - A. Mass transfer is accompanied by reaction
 - B. Diffusion of A is in stagnant medium B
 - C. Molecular mean free path is high
 - D. There is equimolar counter diffusion

Ans X 1. A

- X 2. B
- X 3. C
- √ 4. D

Question ID: 5834939945

- **Q.32** The molar composition of a gas is $10\%~H_2$, $10\%~O_2$, $30\%~CO_2$ and balance H_2O . If 50
 - % H_2O condenses, the final mole percent of H_2 in the dry gas will be
 - A. 10%
 - B. 5%
 - C. 18.18%
 - D. 20%

- Ans X 1. A
 - X 2. B
 - X 3. C
 - √ 4. D

Q.33		
	The vessel dispersion number of an ideal CSTR is	
	A1	
	B. 0	
	C. 1	
	D. ∞	
Ans	★ 1. A	
	★ 2. B	
	X 3. C	
	✓ 4. D	
		Question ID : 5834939950
Q.34	A counter flow heat exchanger is used to heat water from 20 °C to	o 80 °C by using
	exhaust gas entering at 140 °C and leaving at 80 °C. What is the LMTI	O for the best heat
	exchanger?	
	A. 80 °C	
	B. 60 °C	
	C. 110 °C	
	D. Not determinable	
Ans	★ 1. A	
	X 2. B	
	※ 3. C	
	✓ 4. D	
		2 11 12
		Question ID : 5834939923
Q.35	If P_A is the partial pressure of Vapour A in a vapour-gas mixture and P_S is the	e saturation vapour
	pressure of liquid A, condensation of vapour-gas just begins when	
	$A. P_A > P_S$	
	B. $P_A < P_S$	
	C. $P_A = P_S$	
	D. None of above	
Ans	X 1. A	
	X 2. B	
	✓ 3. C	
	★ 4. D	
		Question ID : 5834939949
		GACCHOILID . VOUTUUUTU

Q.36	Vulcanisati	on of mibbo	

- A. Decreases in tensile strength
- B. Increases its ozone and oxygen reactivity
- C. Increases its oil and solvent resistance
- D. Converts its plasticity to elasticity

Ans X 1. A

- X 2. B
- X 3. C
- √ 4. D

Question ID: 5834939961

Q.37 For a certain gas phase reaction, $-\frac{dC_A}{dt} = K C_A^n$. The rate of reaction in terms of partial pressures may be expressed as

- A. $-\frac{dP_A}{dt} = K P_A^n$
- B. $-\frac{dP_A}{dt} = KRT P_A^n$
- C. $-\frac{dP_A}{dt} = K (RT)^{1-n} P_A^n$
- D. $-\frac{dP_A}{dt} = K (RT)^n P_A^{1-n}$

Ans X 1. A

- 🗶 2. B
- √ 3. C
- X 4. D

Question ID: 5834939951

Q.38 ____ explains the equilibrium constant for any chemical reaction.

- A. Henry's law
- B. Law of mass action
- C. Hess's law
- D. Raoult's law

Ans X 1. A

- ✓ 2. B
- X 3. C
- X 4. D

Q.39	For the same feed, feed quality and separation (in a distillation column) with increase of total pressure, number of ideal plates will	
	A. Increase	
	B. Decrease	
	C. Remains same	
	D. Data insufficient, can't be predicted	
Ans		
	★ 2. B	
	★ 3. C	
	★ 4. D	
	V 4. D	
		Question ID : 5834939946
Q.40	The ratio of surface convection resistance to internal conduction resi	istance is called
	A. Grashof number	
	B. Biot number	
	C. Stanton number	
	D. Prandtl number	
Ans	★ 1. A	
	✓ 2. B	
	X 3. C	
	★ 4. D	
		Question ID : 5834939922
Q.41	According to Debye-Bueche theory, the viscosity of a polymer solu	tion or melts is
	proportional to	
	A. Concentration	
	B. Molecular weight	
	C. Both (A) and (B)	
	D. None of these	
Ans		
	★ 2. B	
	✓ 3. C	
	X 4. D	
		Question ID : 5834939962

Q.42	The value of Vanderwaals' constant 'a' for gases O_2 , N_2 , NH_3 and	CH ₄ are 1.36, 1.39,
	4.17 and 2.25 l^{-1} .atm.mol ⁻² respectively. The gas which can be most	
	A. O ₂	
	B. N ₂	
	C. <i>NH</i> ₃	
	D. <i>CH</i> ₄	
Ans	★ 1. A	
	X 2. B	
	✓ 3. C	
	X 4. D	
		Question ID : 5834939963
Q.43	Buna-N is a polymer of	
	A. Butadiene	
	B. Butadiene and Styrene	
	C. Butadiene and Acrylonitrile	
	D. Styrene and Acrylonitrile	
Ans	★ 1. A	
	★ 2. B	
	✓ 3. C	
	★ 4. D	
		Ougstion ID v 5934030050
		Question ID : 5834939960
Q.44	Entropy is a/an	
	A. State function	
	B. Macroscopic property	
	C. Extensive property	
	D. None of these	
Ans	★ 1. A	
	★ 2. B	
	★ 3. C	
	✓ 4. D	
		0 11 12 22 22 22
		Question ID : 5834939927

5/l chemostat is fed fresh medium at 0.2 l/min having a substrate of consumption in (g/l/min) of the substrate in the reactor is 1.2 0.5 0.9 1.5 1.A 2.B 3.C 4.D	Question ID : 5834939956
1.2 0.5 0.9 1.5 1.A 2.B 3.C 4.D Which of the following solutions will follow Raoult's Law most closely A solution of Benzene, Toluene and O-xylene	
0.5 0.9 1.5 1.A 2.B 3.C 4.D Which of the following solutions will follow Raoult's Law most closely A solution of Benzene, Toluene and O-xylene	
0.9 1.5 1. A 2. B 3. C 4. D Which of the following solutions will follow Raoult's Law most closely A solution of Benzene, Toluene and O-xylene	
1.5 1.A 2.B 3.C 4.D Which of the following solutions will follow Raoult's Law most closely A solution of Benzene, Toluene and O-xylene	
 ✓ 1. A ✓ 2. B ✓ 3. C ✓ 4. D Which of the following solutions will follow Raoult's Law most closely a solution of Benzene, Toluene and O-xylene	
 2. B 3. C 4. D Which of the following solutions will follow Raoult's Law most closely A solution of Benzene, Toluene and O-xylene 	
 ✓ 3. C ✓ 4. D ✓ which of the following solutions will follow Raoult's Law most closely a solution of Benzene, Toluene and O-xylene 	
✓ 4. D Which of the following solutions will follow Raoult's Law most closely A solution of Benzene, Toluene and O-xylene	
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A solution of Benzene, Toluene and O-xylene	
A solution of Benzene, Toluene and O-xylene	
A solution of Benzene, Toluene and O-xylene	?
A solution of Benzene, Toluene and O-xylene	
. 35% solution Camphor in Water	
T .	
. 35% solution NH ₃ in Water	
A Solution of polar organic compounds (Not of homologs of a series)	
∕ 1. A	
√ 2. B	
√ 3. C	
√ 4. D	
	Question ID : 5834939947
Ic-leod gauge is a	
. U tube filled with mercury	
. U tube filled with low vapour pressure oil	
. None of these	
√ 1. A	
₹ 2. B	
√ 3. C	
▼ 4. D	
× ×	U tube filled with mercury U tube filled with low vapour pressure oil Liquid manometer along with pressure amplifier None of these 1. A 2. B 3. C

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Q.48	A solid coated on the surface of a flat plate dissolves in a liquid flowing angle of incidence, Diffusion of the solute occurs through the boundary plate. In such a case, the thickness of the momentum boundary layer is exthan the concentration boundary layer, if Schmidt Number, S_C is $A. > 1$	layer formed on the
	B. <1	
	C. 1	
	D. 0.005	
Ans	√ 1. A	
	★ 2. B	
	X 3. C	
	X 4. D	
		Question ID : 5834939948
Q.49	At what percentage (by volume) of Alcohol composition it forms an aze	otrope with water
	A. 90	
	B. 91.5	
	C. 95	
	D. 99	
Ans	★ 1. A	
	★ 2. B	
	✓ 3. C	
	★ 4. D	
		Question ID : 5834939943
Q.50		
4.00	The conversion for a first order liquid phase reaction $A \rightarrow B$, in a	
	another CSTR of the same volume is connected in series, the	en the percentage
	conversion at the exit of the second reactor will be	
	A. 60 B. 75	
	C. 90	
	D. 100	
Ans	X 1. A	
	✓ 2. B	
	★ 3. C	
	★ 4. D	
		Ougation ID : F9340300F4
		Question ID : 5834939954